Validated Chromatographic Methods for the Simultaneous Determination of Sodium Cromoglycate and Oxymetazoline Hydrochloride in a Combined Dosage Form

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Abstract:
Two chromatographic methods were developed and validated for the simultaneous determination of Sodium Cromoglycate (SCG) and Oxymetazoline Hydrochloride (OXMT). SCG and OXMT are administered in combination for effective treatment of nasal congestion and allergy. The first chromatographic method was based on using aluminum TLC plates pre-coated with silica gel GF254 as the stationary phase and chloroform: methanol: toluene: triethylamine (5: 2: 4:1, by volume) as the mobile phase followed by densitometric measurement of the separated bands at 235 nm. The second method is a high performance liquid chromatographic method for separation and determination of SCG and OXMT using reversed phase C18 column with isocratic elution. The mobile phase composed of acetonitrile: methanol (2: 1, v/v) at flow rate of 1.0 mL/min. Quantitation was achieved with UV detection at 220 nm. The validity of the proposed methods was assessed using the standard addition technique. The obtained results were statistically compared with those obtained by the official methods, showing no significant difference with respect to accuracy and precision at p = 0.05.

Keywords: Sodium cromoglycate; Oxymetazoline hydrochloride; TLC-densitometry; high performance liquid chromatography.
Introduction

Chemically, sodium cromoglycate (SCG) is 5,5′-(2-hydroxypropane-1,3-diyl)bis(oxy)bis(4-oxo-4H-chromene-2-carboxylic acid) disodium salt (Fig. 1a). It exerts its action via preventing the release of mediators that would normally attract inflammatory cells. [1] Oxymetazoline hydrochloride (OXMT) is chemically designated as 3-(4,5-dihydro-1H-imidazol-2-ylmethyl)-2,4-dimethyl-6-tert-ethyl-phenol hydrochloride salt (Fig. 1b). It is a sympathomimetic agent that selectively acts on α1 and partially on α2 adrenergic receptors. [2] Both drugs are co-formulated in a nasal spray dosage form and are widely used for effective treatment of nasal congestion and allergy. Several methods have been reported for the determination of SCG in pharmaceutical preparations such as spectrophotometry [3, 4], electrophoresis [5], electrochemical [6, 7] and HPLC methods. [8-14] OXMT has also been analyzed by several methods in pharmaceutical preparations such as spectrophotometry [3, 15-23], capillary electrophoresis [24, 25], gas chromatography [26, 27] and HPLC. [28-35] Abdel-Aziz et al. [3] developed new accurate, sensitive and selective spectrophotometric and spectrofluorimetric methods for determination of SCG and OXMT. To the best of our knowledge, no reported chromatographic methods were found for simultaneous determination of the cited drugs in their combination. The aim of this work is to develop simple chromatographic methods for the simultaneous determination of SCG and OXMT in pharmaceutical dosage form.

Experimental

2.1. Instruments

The thin-layer chromatographic (TLC) system consisted of a Camag Linomat autosampler (Muttenz, Switzerland), a Camag microsyringe (100 mL) and a Camag 35/N/30319 TLC scanner with win CATS software; an ultraviolet (UV) lamp with a short wavelength at 254 nm (Desaga, Wiesloch, Germany); and TLC plates precoated with silica gel G,F, 10×20 cm, 0.25 mm thickness (Merck, Darmstadt, Germany).

HPLC system consisted of Agilent 1100 series liquid chromatograph, Waldborn, Germany. It consists of Isocratic pump, model G 1310 A, UV detector—Model G 1314 A, A Rheodyne injector—Model 7225/77251, (Rohnert park, CA, USA) Equipped with a prominence Autosampler (model SIL-20A). C18, Zorbax Eclipse XDB (4.6 x 150 mm), particle size (5 μm), USA.

2.2. Materials and reagents

2.2.1. Pure standard

SCG pure sample was supplied by SIGMA Pharmaceutical industries, Egypt. Pure sample of OXMT was supplied by National Organization for Drug Control and Research (NODCAR), Egypt. Their purity were checked and found to be 101.5 ± 1.01 and 99.93 ± 1.01 for SCG and OXMT, respectively, according to the USP reference methods [36] which are spectrophotometric and HPLC methods for SCG and OXMT, respectively.

2.2.2. Pharmaceutical dosage form

Nasocrom® nasal spray (SIGMA Pharmaceutical industries, Egypt), labeled to contain 2g of SCG and 0.025g of OXMT per 100 mL (Batch number: 11087) was obtained from the local market.

2.2.3. Chemicals and reagents

All chemicals used throughout the work were of analytical grade and solvents for HPLC were of HPLC grade. These included methanol, chloroform, triethylamine and acetonitrile (Sigma-Aldrich, Belgium), and double distilled deionized water (Otsuka, Cairo, Egypt). Toluene was purchased from Al-Nasr Pharmaceutical Chemicals Co., Cairo, Egypt.

2.2.4. Standard solutions

Standard stock solution of each of SCG and OXMT: 1.0 mg/mL in methanol (for TLC method) and 10 mg/mL of SCG and 1.0 mg/mL OXMT in mobile phase (for HPLC method).

2.2.5. Working Solutions

For HPLC method. Working solutions of SCG (1.0 μg/mL) and OXMT (100μg/mL) were prepared from their respective stock solutions using mobile phase as a solvent.

2.3. Procedures

2.3.1. Construction of the calibration curves

For TLC-spectrodensitometric method. Accurately measured aliquots of SCG and OXMT stock standard solution (1mg/mL) were spotted onto two different TLC plates using Camag Linomat autosampler with microsyringe (100 μL). The plates were then developed by the ascending technique using chloroform: methanol: toluene: triethylamine (5: 2: 4:1, by volume) as a mobile phase. The plates were then removed and air-dried. The chromatogram was scanned at 235 nm. Calibration curves representing the relationship
between integrated peak area and the corresponding concentrations of each of SCG (0.4-4.4 µg/band) and OXMT (0.5-20.0 µg/band) were plotted.

2.3.2. For HPLC method:

Aliquots of SCG and OXMT equivalent to 900-8000 µg and 100-900 µg, respectively, where separately and accurately transferred each from its respective working standard solution (1.0 mg/mL SCG and 0.1 mg/mL OXMT) into three separate sets of 10-mL measuring flasks and the volume was then completed to the mark with the mobile phase (acetonitrile: methanol (2: 1, v/v)). A 20-µL aliquot of each solution was injected into an C18, Zorbax Eclipse XDB (4.6 x 150 mm), particle size (5 µm), using the mobile phase, at flow rate 1.0 mL/min and UV detection at 220 nm. Two calibration curves were constructed by plotting the relative peak area, using an external standard of 400 µg/mL SCG and 50 µg/mL OXMT, against the corresponding concentrations of each drug.

2.3.3. Application to pharmaceutical formulations

Appropriate dilutions from Nasocrom® nasal drops were made using methanol to obtain solutions having concentrations within the linearity range of each of SCG and OXMT, these dilutions were analyzed by the proposed chromatographic methods as detailed under linearity. The concentrations of SCG and OXMT were obtained from the computed regression equations and the validity of the methods was further assessed by applying the standard addition technique.

**Results**

Several trials were conducted to develop the optimum chromatographic conditions for the sufficient separation of both drugs. The results of the TLC system were satisfactory when using chloroform: methanol: toluene: triethylamine (5: 2: 4: 1, by volume) as the mobile phase. RF values were found to be 0.1±0.01 and 0.3±0.01 for SCG and OXMT, respectively as shown in (Fig. 2). This separation allows the determination of SCG and OXMT at 235 nm without any interference from each other.

HPLC method was also tried to separate SCG and OXMT, therefore several trials have been undertaken to reach the optimum stationary/mobile phases matching. Good chromatographic separation of the two drugs in their binary mixtures could be achieved by using an C18, Zorbax Eclipse XDB (4.6 × 150 mm), particle size (5 µm), with a mobile phase consisting of (acetonitrile: methanol (2: 1, v/v)) at flow rate 1.0 mL/min, followed by UV detection at 220 nm, (Fig. 3). Calibration was performed for both TLC and HPLC for the determination of the cited drugs. An overall system suitability testing was calculated (Table I) to determine whether the operating system performed properly. The proposed methods were validated according to International Conference on Harmonization (ICH) guidelines (Table II). The table also shows the assay parameters of the regression equations and the ranges of concentration.

The proposed methods were successfully applied for determination of SCG and OXMT in Nasocrom® nasal spray. The results shown in table (III) were satisfactory. The validity of the proposed methods was assessed by applying the standard addition technique, no interference due to excipients was observed as shown from the results in table (III).

The results obtained by applying the proposed methods for analysis of pure SCG and OXMT compared to those obtained by applying the official methods [36]; they showed no significance difference regarding accuracy and precision table (IV).

**Discussion**

Planar chromatography with precise application of the samples and computer controlled evaluation and quantification of the developed chromatograms has been considered to be a reliable technique for purity control and for quantitative drug testing [37]. Therefore the aim of this work is to develop simple, accurate, rapid, specific and valid spectro-densitometric method. This separation allows the determination of SCG and OXMT without any interference from each other. Linear relationship was found to exist between the integrated peak area of the separated bands at the selected wavelength (235 nm) and the corresponding concentration of SCG in the range of 0.4-4.4 µg/band and a polynomial one was found in the range of 0.5-20.0 µg/band for OXMT, figures (19, 20).

The regression equations were computed and found to be:

\[ A = 0.375X + 0.309 \]  \[ r = 0.9999 \]  for SCG

\[ A = -0.003X + 0.219X + 0.282 \]  \[ r = 0.9990 \]  for OXMT

Where, \( A \) is the integrated peak area x 10-4, \( X \) is the concentration in µg/band and \( r \) is the correlation coefficient.

The suggested chromatographic system for the HPLC method allows complete base line separation at reasonable time. The linearity of the detector's response of the studied drugs was determined by plotting relative peak area (calculated following the external standard technique using 400 µg/mL SCG and 50 µg/mL OXMT as the external standards for SCG and OXMT, respectively) versus concentrations and linear correlation was obtained. The regression equations were computed for SCG and OXMT and found to be:

\[ A = 0.002X + 0.059 \]  \[ r = 1.0000 \]  for SCG

\[ A = 0.020X + 0.010 \]  \[ r = 0.9995 \]  for OXMT

Where, \( A \) is the relative peak area, \( X \) is the drug concentration in µg/mL and \( r \) is the correlation coefficient.
2.4. Method Validation

The proposed methods were validated according to the ICH Q2 (R1) recommendations [38]. The method was validated for parameters such as system suitability, linearity, limit of detection (LOD), limit of quantitation (LOQ), accuracy, precision and selectivity.

2.4.1. System suitability

The system suitability test is an integral part of chromatographic method development and it is used to verify that the system is adequate for the analysis to be performed; the parameters Q8 for SCG and OXMT were evaluated. The suitability of the chromatographic system was determined according to USP guidelines and with acceptance of the obtained parameter values. [36]

2.4.2. Linearity and ranges

Under the above mentioned experimental conditions, linear relationships were obtained by plotting the drug concentrations either against integrated peak areas or relative peak areas for each drug, for TLC and HPLC methods, respectively.

2.4.3. Accuracy

The accuracy of the proposed methods was validated by analyzing pure samples of each SCG and OXMT in triplicate. The concentrations of the active drugs were calculated from the corresponding regression equations.

2.4.4. Precision

It was evaluated by calculating intra and inter-day precision. By repeating the assay of three different concentrations of each of the cited drugs three times in the same day and assaying the same samples in triplicate on three successive days using the developed chromatographic methods. RSD values were then calculated for each sample.

2.4.5. Specificity

The specificity of the developed methods was investigated by application of mixtures containing the two drugs in different ratios to TLC plate and the method under the chromatographic conditions were then followed. The bands of the two drugs in the prepared mixtures were confirmed by comparing their Rf values with that of standard drugs solutions. (in TLC method). Specificity was ascertained by analyzing different mixtures containing SCG and OXMT in different ratios. Parameters such as resolution, capacity and selectivity factors for the separated chromatographic peaks were then calculated (in HPLC method).

Conclusion

The suggested chromatographic methods provide simple, sensitive, accurate and reproducible methods for quantitative analysis of SCG and OXMT in their binary mixtures and pharmaceutical dosage form. The developed TLC method is highly sensitive. It has the advantages of short run time, large sample capacity and use of minimal volume of solvents. HPLC method gives a good resolution between the proposed components within suitable analysis time; it is highly specific but more expensive.

References


**Table I:** Parameters required for system suitability test of TLC-densitometric and HPLC methods.

<table>
<thead>
<tr>
<th>Parameter</th>
<th>TLC-Densitometric method</th>
<th>HPLC method</th>
<th>Reference values</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>SCG</td>
<td>OXMT</td>
<td>SCG</td>
</tr>
<tr>
<td>Retention time ($t_R$) [min]</td>
<td></td>
<td>1.49</td>
<td>2.67</td>
</tr>
<tr>
<td>Retardation factor ($R_f$)</td>
<td>0.1</td>
<td>0.3</td>
<td></td>
</tr>
<tr>
<td>Tailing factor ($T$)</td>
<td>0.9</td>
<td>0.87</td>
<td>0.98</td>
</tr>
<tr>
<td>Capacity factor ($K'$)</td>
<td>9.0</td>
<td>7.0</td>
<td>2.34</td>
</tr>
<tr>
<td>Selectivity factor ($α$)</td>
<td>1.29</td>
<td>3.03</td>
<td></td>
</tr>
<tr>
<td>Resolution factor ($Rs$)</td>
<td>5.14</td>
<td>3.38</td>
<td></td>
</tr>
<tr>
<td>Column efficiency ($N$)</td>
<td></td>
<td></td>
<td>256</td>
</tr>
<tr>
<td>HETP <em>[m]</em></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

HETP * = height equivalent to theoretical plates (length of column in mm/ N).
Table II: Assay validation sheet of the proposed methods for the simultaneous determination of Sodium cromoglycate (SCG) and Oxymetazoline hydrochloride (OXMT).

<table>
<thead>
<tr>
<th>Parameters</th>
<th>TLC-Densitometric method</th>
<th>HPLC method</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>SCG</td>
<td>OXMT</td>
</tr>
<tr>
<td>Range</td>
<td>0.4-4.4 µg/band</td>
<td>0.5-20.0 µg/band</td>
</tr>
<tr>
<td>Slope 1</td>
<td>0.375</td>
<td>-0.003</td>
</tr>
<tr>
<td>Slope 2</td>
<td>-----</td>
<td>0.219</td>
</tr>
<tr>
<td>Intercept</td>
<td>0.309</td>
<td>0.282</td>
</tr>
<tr>
<td>SE of the slope</td>
<td>0.0084</td>
<td>0.0045</td>
</tr>
<tr>
<td>SE of the intercept</td>
<td>0.0232</td>
<td>0.0525</td>
</tr>
<tr>
<td>Correlation coefficient (r)</td>
<td>0.9999</td>
<td>0.9990</td>
</tr>
<tr>
<td>LOD</td>
<td>0.100 (µg/band)</td>
<td>0.150 (µg/band)</td>
</tr>
<tr>
<td>LOQ</td>
<td>0.400 (µg/band)</td>
<td>0.500 (µg/band)</td>
</tr>
<tr>
<td>Accuracy</td>
<td>100.67±1.805</td>
<td>99.63±1.290</td>
</tr>
<tr>
<td>Precision (RSD)</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Repeatability</td>
<td>0.998</td>
<td>1.520</td>
</tr>
<tr>
<td>Intermediate precision</td>
<td>1.709</td>
<td>2.400</td>
</tr>
</tbody>
</table>

*aThe intraday RSD % and *bThe inter-day RSD %, for TLC RSD% of three concentrations (1.50, 2.50 and 3.50 µg/band) of SCG and (3.50, 5.50 and 7.50 µg/band) of OXMT repeated three times within the day. For HPLC RSD% of three concentrations (90.00, 150.00 and 500.00 µg/mL) of SCG and (10.00, 50.00 and 80.00 µg/mL) of OXMT repeated three times within the day.
Table III: Determination of Sodium cromoglycate (SCG) and Oxymetazoline hydrochloride (OXMT) in their Dosage form and application of standard addition technique using the proposed methods.

<table>
<thead>
<tr>
<th>Nasocrom® nasal drops</th>
<th>T</th>
<th>L</th>
<th>C</th>
<th>H</th>
<th>P</th>
<th>L</th>
<th>C</th>
</tr>
</thead>
<tbody>
<tr>
<td>B. N. 80724</td>
<td>SCG</td>
<td>OXMT</td>
<td>SCG</td>
<td>OXMT</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Mean Recovery</td>
<td>± RSD</td>
<td>104.00±1.351</td>
<td>100.83±2.100</td>
<td>103.56±0.630</td>
<td>101.16±0.960</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Standard addition</td>
<td>(mean±RSD)</td>
<td>99.45±1.141</td>
<td>100.73±1.548</td>
<td>99.00±0.006</td>
<td>100.63±1.475</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

aAverage of three determinations of dosage form.
bAverage of three determinations of dosage form after the addition of different amounts of SCG or OXMT pure standard (standard addition technique).

Table IV: Statistical comparison for the results obtained by the proposed methods and the official methods for the analysis of Sodium cromoglycate (SCG) and Oxymetazoline hydrochloride (OXMT).

<table>
<thead>
<tr>
<th>Parameter</th>
<th>T</th>
<th>L</th>
<th>C</th>
<th>H</th>
<th>P</th>
<th>L</th>
<th>C</th>
<th>Official method</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>SCG</td>
<td>OXMT</td>
<td>SCG</td>
<td>OXMT</td>
<td>SCG</td>
<td>OXMT</td>
<td>SCG</td>
<td>OXMT</td>
</tr>
<tr>
<td>Mean</td>
<td>100.67</td>
<td>99.63</td>
<td>100.71</td>
<td>99.65</td>
<td>101.50</td>
<td>99.93</td>
<td></td>
<td></td>
</tr>
<tr>
<td>SD</td>
<td>1.818</td>
<td>1.285</td>
<td>1.664</td>
<td>1.150</td>
<td>1.005</td>
<td>1.050</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Variance</td>
<td>3.303</td>
<td>1.652</td>
<td>2.769</td>
<td>1.322</td>
<td>1.010</td>
<td>1.010</td>
<td></td>
<td></td>
</tr>
<tr>
<td>n</td>
<td>5</td>
<td>5</td>
<td>5</td>
<td>5</td>
<td>5</td>
<td>5</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Student’s t test</td>
<td>0.889</td>
<td>0.410</td>
<td>0.914</td>
<td>0.408</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>(2.3060)*</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>F value</td>
<td>3.271</td>
<td>1.636</td>
<td>2.742</td>
<td>1.309</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>(6.3882)*</td>
<td></td>
<td></td>
<td></td>
<td></td>
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</tbody>
</table>

* These values represent the corresponding tabulated values of t and F at P=0.05
a) Direct absorbance measurement at λmax 326 nm using Sodium phosphate buffer (pH 7.4) as a blank.
b) HPLC method; mobile phase [water: methanol: 1M sodium acetate: glacial acetic acid (46:40: 10:4)] with UV detection at 280 nm
Figure 1: Chemical structures of (a) sodium cromoglycate and (b) oxymetazoline hydrochloride

Fig. 2: TLC chromatogram of mixture of Sodium Cromoglycate (SCG) (RF=0.10±0.01) and Oxymetazoline Hydrochloride (OXMT) (RF=0.30±0.01) at 235 nm using chloroform: methanol: toluene: triethylamine (5: 2: 4:1, by volume) as a developing system.
Fig. 3: HPLC chromatogram of a resolved mixture of Sodium Cromoglycate (SCG) ($t_R=1.49$) and Oxymetazoline Hydrochloride (OXMT) ($t_R=2.67$) using mobile phase of acetonitrile: methanol (2: 1, v/v).